Ph	me <u>Answer K</u> ysics riod		Date arges	Electrostatics WS #2 Mrs. Nadworny
	rections: Read online text ethod and proper signific		. Solve the following probl show ALL work.	lems using the GUESS
1.	What is a correct value for the charge on an electron?			
	(A) 1.60 x 10 ⁻¹² μC	(B) 1.60 x 10 ⁻¹⁵ mC	(C) 1.60 x 10 ⁻²² kC	(D) 1.60 x 10 ⁻²⁴ MC
2.	An object with +10 elementary charges is grounded and becomes neutral. What is the best explanation for this occurrence? (A) The object gained 10 protons from the ground (B) The object gained 10 electrons from the ground (C) The object lost 10 protons to the ground (D) The object lost 10 electrons to the ground			
3.	An object cannot have a charge of			
	(A) 3.2 x 10 ⁻¹⁹ C	(B) 4.5 x 10 ⁻¹⁹ C	(C) 8.0 x 10 ⁻¹⁹ C	(D) 9.6 x 10 ⁻¹⁹ C
4.	The charge to mass ratio of an electron is			
	(A) $1.76 \times 10^{-11} \text{ C/kg}$	(B) $5.69 \times 10^{-12} \text{C/kg}$	(C) $1.76 \times 10^{11} \text{ C/kg}$	(D) $5.69 \times 10^{12} \text{C/kg}$
5.	Which quantity of excess electric charge could be found on an object?			
	(A) 6.25 x 10 ⁻¹⁹ C (B) 4.80 x 10 ⁻¹⁹ C		(C) 6.25 elementary char (D) 1.60 elementary char	_
6.	A rubber rod becomes negatively charged when it is rubbed with fur. The net negative charge accumulates because the rubber rod			
	(A) gains electrons	(B) loses protons	(C) gains protons	(D) loses electrons
7.	What is the smallest electric charge that can be put on an object?			
	(A) 9.11 x 10 ⁻³¹ C (B) 1.60 x 10 ⁻¹⁹ C		(C) 9.00 x 10 ⁹ C (D) 6.25 x 10 ¹⁸ C	
8.	Compared to the charge on a proton, the charge on an electron has the			
	(A) same sign and a smaller magnitude (C		(C) opposite sign and the same magnitude	

(B) opposite sign and a small magnitude

(A) more free electrons

(B) fewer free electrons

9. Compared to an insulator, a conductor of electric current has

(D) same sign and the same magnitude

(C) more free atoms

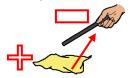
(D) fewer free atoms

10. Show the charge flow for each situation below. Show the charge on each AFTER.

a. A balloon is rubbed against a Northport physics student's hair. Electrons flow from the

hair to the balloon.





c. A piece of cotton is rubbed against a piece of acetate. The acetate ends up positive.



11. In the following problems, convert between elementary charges and coulombs.

a. What is the charge of four electrons in Coulombs?

$$(4e^{-}) \left(\frac{-1.60 \times 10^{-19} C}{1e^{-}} \right) = -6.40 \times 10^{-19} C$$

b. What is the charge of four protons in Coulombs?

$$(4e)\left(\frac{1.60\times10^{-19}\,\mathrm{C}}{1e}\right) = 6.40\times10^{-19}\,\mathrm{C}$$

c. How many elementary charges are in 5.76 x 10⁻¹⁶ C? Are they electrons or protons?

$$(5.76 \times 10^{-19} \text{ C}) \left(\frac{1\text{e}}{1.60 \times 10^{-19} \text{ C}} \right) = 3.60 \times 10^3 \text{ protons}$$

d. How many elementary charges are in -4.3 x 10⁻⁶ C? Are they electrons or protons?

$$(-4.3\times10^{-6} \, \text{C}) \left(\frac{1\text{e}^{-}}{-1.60\times10^{-19} \, \text{C}}\right) = 2.7\times10^{13} \text{ electrons}$$

Answers in size order: 6.40×10^{-19} , 6.40×10^{-19} , 3.60×10^{3} , 2.7×10^{13}